

# Appendix A

## AP BIOLOGY EQUATIONS AND FORMULAS

STATISTICAL ANALYSIS AND PROBABILITY									
Standard Error			Mean						
$SE_{\bar{x}} = \frac{s}{\sqrt{n}}$			$\bar{x} = \frac{1}{n} \sum_{i=1}^n x_i$						
Standard Deviation			Chi-Square						
$s = \sqrt{\frac{\sum (x_i - \bar{x})^2}{n - 1}}$			$\chi^2 = \sum \frac{(o - e)^2}{e}$						
CHI-SQUARE TABLE									
Degrees of Freedom									
p	1	2	3	4	5	6	7	8	
0.05	3.84	5.99	7.82	9.49	11.07	12.59	14.07	15.51	
0.01	6.64	9.32	11.34	13.28	15.09	16.81	18.48	20.09	
LAWS OF PROBABILITY									
If A and B are mutually exclusive, then P (A or B) = P(A) + P(B)									
If A and B are independent, then P (A and B) = P(A) x P(B)									
HARDY-WEINBERG EQUATIONS									
$p^2 + 2pq + q^2 = 1$			$p$ = frequency of the dominant allele in a population						
$p + q = 1$			$q$ = frequency of the recessive allele in a population						
METRIC PREFIXES									
Factor		Prefix			Symbol				
10 <sup>9</sup>		giga			G				
10 <sup>6</sup>		mega			M				
10 <sup>3</sup>		kilo			k				
10 <sup>-2</sup>		centi			c				
10 <sup>-3</sup>		milli			m				
10 <sup>-6</sup>		micro			μ				
10 <sup>-9</sup>		nano			n				
10 <sup>-12</sup>		pico			p				
Mode = value that occurs most frequently in a data set									
Median = middle value that separates the greater and lesser halves of a data set									
Mean = sum of all data points divided by number of data points									
Range = value obtained by subtracting the smallest observation (sample minimum) from the greatest (sample maximum)									

RATE AND GROWTH		Water Potential ( $\Psi$ )
<b>Rate</b> $dY/dt$ <b>Population Growth</b> $dN/dt = B - D$ <b>Exponential Growth</b> $\frac{dN}{dt} = r_{\max} N$ <b>Logistic Growth</b> $\frac{dN}{dt} = r_{\max} N \left( \frac{K - N}{K} \right)$	$dY$ = amount of change $t$ = time $B$ = birth rate $D$ = death rate $N$ = population size $K$ = carrying capacity $r_{\max}$ = maximum per capita growth rate of population	$\Psi = \Psi_p + \Psi_s$ $\Psi_p$ = pressure potential $\Psi_s$ = solute potential The water potential will be equal to the solute potential of a solution in an open container, since the pressure potential of the solution in an open container is zero. <b>The Solute Potential of the Solution</b> $\Psi_s = -iCRT$ $i$ = ionization constant (For sucrose this is 1.0 because sucrose does not ionize in water.) $C$ = molar concentration $R$ = pressure constant ( $R = 0.0831$ liter bars/mole K) $T$ = temperature in Kelvin ( $273 + ^\circ\text{C}$ )
<b>Temperature Coefficient <math>Q_{10}</math></b> $Q_{10} = \left( \frac{k_2}{k_1} \right)^{\frac{10}{t_2 - t_1}}$ <b>Primary Productivity Calculation</b> $\text{mg O}_2/\text{L} \times 0.698 = \text{mL O}_2/\text{L}$ $\text{mL O}_2/\text{L} \times 0.536 = \text{mg carbon fixed/L}$	$t_2$ = higher temperature $t_1$ = lower temperature $k_2$ = metabolic rate at $t_2$ $k_1$ = metabolic rate at $t_1$ $Q_{10}$ = the <i>factor</i> by which the reaction rate increases when the temperature is raised by ten degrees	
SURFACE AREA AND VOLUME		Dilution – used to create a dilute solution from a concentrated stock solution
<b>Volume of a Sphere</b> $V = 4/3 \pi r^3$ <b>Volume of a Cube (or Square Column)</b> $V = l w h$ <b>Volume of a Column</b> $V = \pi r^2 h$ <b>Surface Area of a Sphere</b> $A = 4 \pi r^2$ <b>Surface Area of a Cube</b> $A = 6 a$ <b>Surface Area of a Rectangular Solid</b> $A = \Sigma$ (surface area of each side)	$r$ = radius $l$ = length $h$ = height $w$ = width $A$ = surface area $V$ = volume $\Sigma$ = Sum of all $a$ = surface area of one side of the cube	$C_i V_i = C_f V_f$ $i$ = initial (starting) $C$ = concentration of solute $f$ = final (desired) $V$ = volume of solution <b>Gibbs Free Energy</b> $\Delta G = \Delta H - T \Delta S$ $\Delta G$ = change in Gibbs free energy $\Delta S$ = change in entropy $\Delta H$ = change in enthalpy $T$ = absolute temperature (in Kelvin) <b>pH</b> = $-\log [H^+]$